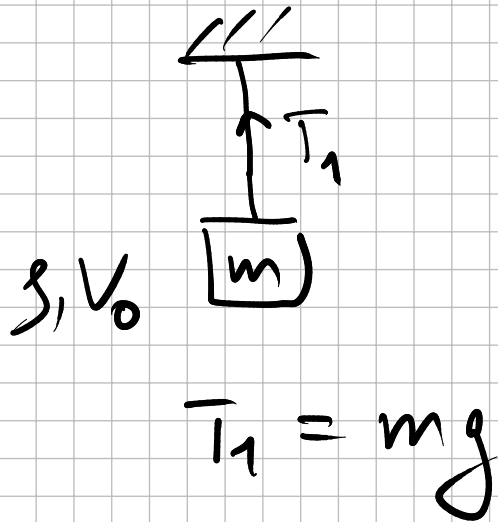


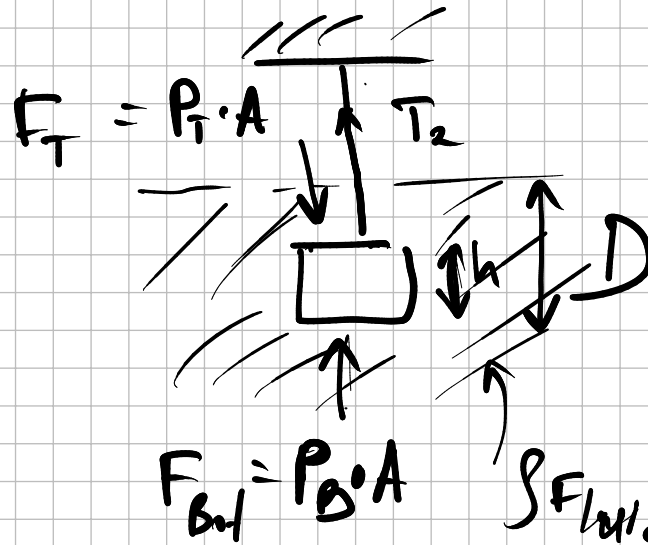
Buoyancy



$$T_1 = mg$$

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$$\frac{T_1 - T_2}{T_1} = \frac{\rho \cancel{F} \cdot \cancel{g} h A}{\rho \cdot \cancel{g} \cdot \cancel{V_0}}$$



$$F_{\text{bu}} = P_B \cdot A \quad \rho_F \cdot g \cdot D \cdot A$$

$$T_2 - P_T \cdot A - P_{\text{bot}} \cdot A - mg = 0$$

$$T_2 = mg - \underbrace{(P_{\text{bot}} - P_T) \cdot A}_{F_{\text{Buoyancy}}}$$

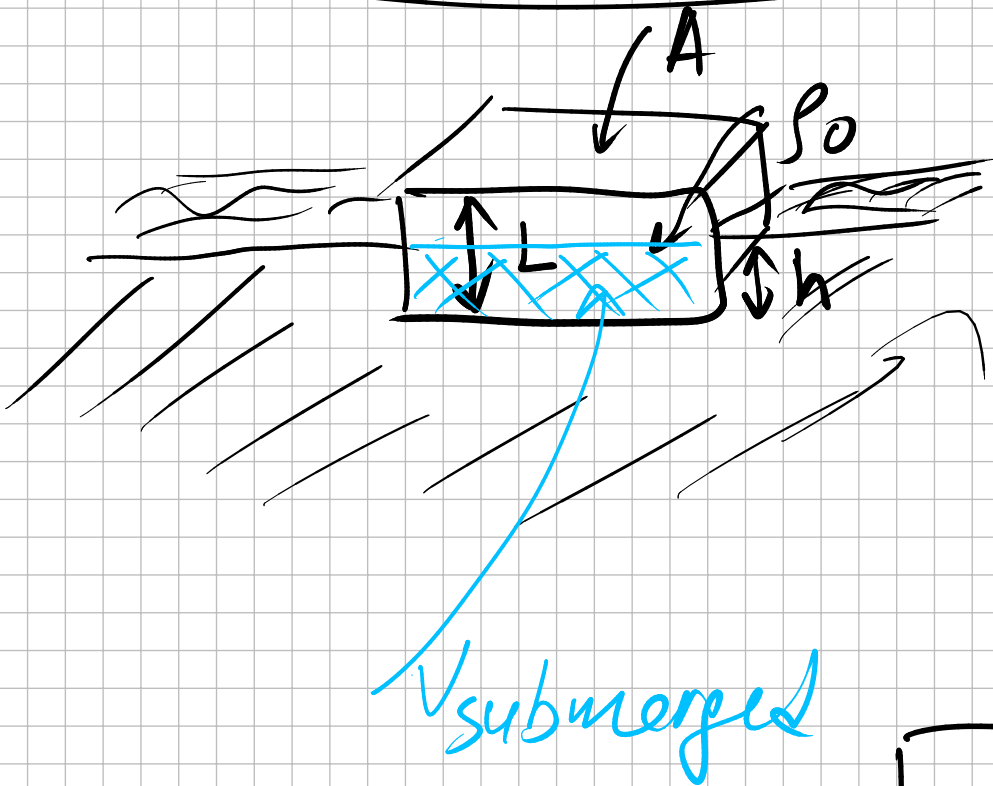
$$= mg - \rho_F g h \cdot A$$

$$P_T = \rho_F g (D - h)$$

$$P_B = \rho_F \cdot g \cdot D$$

$$\rho_0 = \rho_F \frac{T_1}{T_1 - T_2}$$

Partially Submerged case

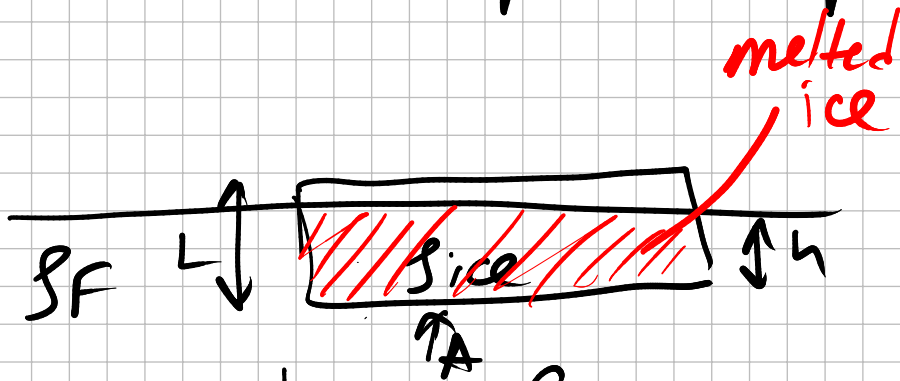


$$\begin{aligned} mg &= F_{\text{Buoyancy}} \\ &= \rho_F \cdot g \cdot \underbrace{V_{\text{submerged}}}_{A \cdot h} \end{aligned}$$

$$mg = \rho_0 \underbrace{V_{\text{object}}}_{A \cdot L} \cdot g$$

$$\frac{\rho_0}{\rho_F} = \frac{h}{L}$$

Ice pack problem



$$\rho_{ice} = 917 \text{ kg/m}^3$$

$$\rho_F = 1000 \text{ kg/m}^3$$

$$\frac{h_{ice}}{L} = \frac{\rho_{ice}}{\rho_F} = \frac{917}{1000} \approx 0.92$$

$$m_{ice} = \rho_{ice} \cdot V_{ice} = \rho_{water} \cdot V_{water}$$

$$\Rightarrow \frac{h_{ice}}{L} = \frac{\rho_{ice}}{\rho_{water}} = \frac{V_{water}}{V_{ice}} = \frac{\cancel{A} \cdot h_{melt\ water}}{A \cdot L}$$