

Phys 630 S'24 Problem Set 2 Solutions

Kardar, Chapter 1

$$6.a) dE = TdS + \underbrace{\mu dN}$$

Chemical work in converting dN liquid quartz molecules to crystal or glass

$$S = S(E, N), \quad dS = \frac{dE}{T} - \frac{\mu}{T} dN = \left. \frac{\partial S}{\partial E} \right|_N dE + \left. \frac{\partial S}{\partial N} \right|_E dN$$

$$\Rightarrow \left. \frac{\partial S}{\partial E} \right|_N = \frac{1}{T}, \quad \left. \frac{\partial S}{\partial N} \right|_E = -\frac{\mu}{T}$$

$$\text{Extensivity: } S(\lambda E, \lambda N) = \lambda S(E, N)$$

$$\left. \frac{\partial}{\partial \lambda} \right|_{\lambda=1} : \left. \frac{\partial S}{\partial E} \right|_N \cdot E + \left. \frac{\partial S}{\partial N} \right|_E \cdot N = S$$

$$S = \frac{1}{T} E - \frac{\mu}{T} N$$

$$\Rightarrow \mu = \frac{E - TS}{N} \quad (\text{Different for crystal and glass})$$

$$\downarrow) C_x = \alpha T^2, \quad C_g = \beta T$$

crystal

glass

$$S = \int \frac{dQ}{T} = \int_0^T \frac{C_V dT'}{T'} = \begin{cases} \int_0^T \alpha T'^2 dT' = \frac{\alpha}{3} T^3 & \text{crystal} \\ \int_0^T \beta dT' = \beta T & \text{glass} \end{cases}$$

$$c) dE = T dS = C_V dT, \quad E(T=0) = E_0$$

$$= \begin{cases} \alpha T^3 dT & \text{crystal} \\ \beta T & \text{glass} \end{cases}$$

$$E = \begin{cases} E_0 + \int_0^T \alpha T'^3 dT' = E_0 + \frac{\alpha T^4}{4} & \text{crystal} \\ E_0 + \int_0^T \beta T' dT' = E_0 + \frac{\beta T^2}{2} & \text{glass} \end{cases}$$

d) Chemical Equilibrium between crystal and glass phases:

$$\mu_x = \mu_g \quad \text{at melting transition}$$

$$\frac{1}{N} (E - TS) \Big|_x = \frac{1}{N} (E - TS) \Big|_g$$

$$\frac{1}{N} \left(E_0 + \frac{\alpha T^4}{4} - T \cdot \frac{\alpha T^3}{3} \right) = \frac{1}{N} \left(E_0 + \frac{\beta T^2}{2} - T \cdot \beta T \right)$$

$$\Rightarrow -\frac{\alpha}{12} T^4 = -\frac{\beta}{2} T^2$$

$$\Rightarrow T_{\text{melt}} = \sqrt{\frac{6\beta}{\alpha}}$$

e) Latent heat for liquid to crystal transition (or glass)

$$\begin{aligned} L &= (E_{\text{glass}} - E_{\text{crystal}}) \Big|_{T_{\text{melt}}} \\ &= T_{\text{melt}} (S_G - S_X) + N(\cancel{\mu_G} - \cancel{\mu_X}) \Big|_{T_{\text{melt}}} \\ &= T_{\text{melt}} \left(\beta T_{\text{melt}} - \frac{\alpha}{2} T_{\text{melt}}^2 \right) \end{aligned}$$

$$= \beta \cdot \left(\frac{6\beta}{\alpha} \right) - \frac{\alpha}{2} \left(\frac{6\beta}{\alpha} \right)^2$$

$$L = - \frac{6\beta^2}{\alpha}$$

(< 0 ???!!)

↓
Glasses don't satisfy $S(T=0) = 0$,
So we found a nonphysical result by assuming the 3rd law.

8. a) We want a Maxwell relation for $\frac{\partial S}{\partial V}|_{T,N}$, so we need a relation between differentials dV , dT , dN .

Fix $N \rightarrow dN=0$

Consider $F = E - TS$

$$dF = -SdT - PdV$$

$$= \frac{\partial F}{\partial T}|_{V,N} dT + \frac{\partial F}{\partial V}|_{T,N} dV$$

$$\Rightarrow \boxed{\frac{\partial S}{\partial V}|_{T,N} = \frac{\partial P}{\partial T}|_{V,N}}$$

b) $dE = TdS - PdV$

$\rightarrow E$ is naturally a fn. of S and V .

We want to consider E as a fn. of T and V .

$$dE = T dS - P dV \quad (\text{constant } N)$$

$$= T \left(\frac{\partial S}{\partial T} \Big|_V dT + \frac{\partial S}{\partial V} \Big|_T dV \right) - P dV$$

(part a)

$$= T \left(\frac{\partial S}{\partial T} \Big|_V dT + \frac{\partial P}{\partial T} \Big|_V dV \right) - P dV$$

$$= T \frac{\partial S}{\partial T} \Big|_V dT + \left(T \frac{\partial P}{\partial T} \Big|_V - P \right) dV$$

Equation of state: $P(V - Nb) = Nk_B T$

$$\rightarrow \frac{\partial P}{\partial T} \Big|_{V,N} = \frac{Nk_B}{V - Nb} = \frac{P}{T}$$

$$\Rightarrow dE = T \frac{\partial S}{\partial T} \Big|_{V,N} dT \quad \text{for const. } N$$

$$\Rightarrow \frac{\partial E}{\partial V} \Big|_{T,N} = 0$$

$$\rightarrow E(T, V, N) = E(T, N)$$

c) To find C_p and C_v , use

$$dQ = dE + PdV \quad (\text{constant } N)$$

$$= \left(\frac{\partial E}{\partial T} \Big|_V dT + \cancel{\frac{\partial E}{\partial V} \Big|_T dV} \right) + PdV$$

$$= \frac{\partial E}{\partial T} dT + PdV$$

$$C_v = \frac{dQ}{dT} \Big|_V = \frac{\partial E}{\partial T} \Big|_V$$

In terms of T and P ,

$$dQ = \left(\frac{\partial E}{\partial T} \Big|_P dT + \frac{\partial E}{\partial P} \Big|_T dP \right) + P \left(\frac{\partial V}{\partial T} \Big|_P dT + \frac{\partial V}{\partial P} \Big|_T dP \right)$$

$$= \left(\frac{\partial E}{\partial T} + P \frac{\partial V}{\partial T} \Big|_P \right) dT + \left(\frac{\partial E}{\partial P} \Big|_T + \frac{\partial V}{\partial P} \Big|_T \right) dP$$

$$C_p = \frac{dQ}{dT} \Big|_P = \frac{\partial E}{\partial T} \Big|_P + P \frac{\partial V}{\partial T} \Big|_P$$

$$C_p = \left. \frac{\partial E}{\partial T} \right|_V + P \left. \frac{\partial}{\partial T} \left(\frac{Nk_B T}{P} \right) \right|_P \quad \left(\text{because } E=E(T), \right. \\ \left. \frac{\partial E}{\partial T} \Big|_P = \frac{\partial E}{\partial T} \Big|_V \right)$$

$$= C_V + Nk_B$$

$$\gamma \equiv \frac{C_p}{C_V} = \frac{C_V + Nk_B}{C_V} = 1 + \frac{Nk_B}{C_V}$$

$$d) \quad dE = C_V dT = C_V d \left(\frac{P(V - Nb)}{Nk_B} \right)$$

$$= \frac{C_V}{Nk_B} (P dV + (V - Nb) dP)$$

$$\delta Q = dE + P dV = 0 \quad \text{Adiabatic}$$

$$\Rightarrow \frac{C_V}{Nk_B} (P dV + (V - Nb) dP) + P dV = 0$$

$$\frac{dP}{P} = - \left(1 + \frac{Nk_B}{C_V} \right) \frac{dV}{V - Nb}$$

$$\frac{dP}{P} = -\gamma \frac{d(V-Nb)}{V-Nb}$$

$$\ln\left(\frac{P}{P_0}\right) = -\gamma \ln\left(\frac{V-Nb}{V_0}\right)$$

for constants P_0, V_0 .

$$\rightarrow P (V-Nb)^\gamma = \text{constant}$$

9. a) $dQ = TdS = CdT$

$$\rightarrow S_s(T) = \int_0^T \frac{C_s(T')}{T'} dT' = \int_0^T \frac{V\alpha T'^3}{T'} dT'$$

$$S_s = \frac{V\alpha T^3}{3}$$

$$S_n(T) = \int_0^T \frac{C_n(T')}{T'} dT' = \int_0^T \frac{V(\beta T'^3 + \gamma T')}{T'} dT'$$

$$S_n = V\left(\frac{\beta T^3}{3} + \gamma T\right)$$

↳ Latent Heat $L = T_c(S_n(T_c) - S_s(T_c)) = 0$

$$\rightarrow V\left(\frac{\beta T_c^3}{3} + \gamma T_c\right) - \frac{V\alpha T_c^3}{3} = 0$$

$$T_c = \sqrt{\frac{3\gamma}{\alpha - \beta}}$$

($T_c = 0$ solution is due to $S(0) = 0$ by the 3rd Law, and does not represent the phase transition.)

$$c) E_n(T \rightarrow 0) = E_0, \quad E_s(T \rightarrow 0) = E_0 - V\Delta$$

Fix the magnetic field M , # electrons N .

$$dE = TdS = C dT$$

$$E_s(T) = E_s(T \rightarrow 0) + \int_0^T C_s(T') dT'$$

$$= E_0 - V\Delta + \int_0^T V\alpha T'^3 dT'$$

$$= E_0 - V\Delta + \frac{V\alpha T^4}{4}$$

$$E_n(T) = E_n(T \rightarrow 0) + \int_0^T C_n(T') dT'$$

$$= E_0 + \int_0^T V(\beta T'^3 + \gamma T') dT'$$

$$= E_0 + V\left(\frac{\beta T^4}{4} + \frac{\gamma T^2}{2}\right)$$

d) Gibbs Free Energy $G = E - TS - BM = \mu N$, $B = 0$

At the transition temperature T_c , the two phases are in chemical equilibrium $\rightarrow \mu_s = \mu_n$

$$\rightarrow G_s = G_n$$

$$G_s(T_c) = \left(E_0 - V\Delta + \frac{V\alpha T_c^4}{4} \right) - T \frac{V\alpha T_c^3}{3}$$
$$= E_0 - V \left(\Delta + \frac{\alpha T_c^4}{12} \right)$$

$$G_n(T_c) = E_0 + V \left(\frac{\beta T_c^4}{4} + \frac{\delta T_c^2}{2} \right) - T V \left(\frac{\beta T_c^3}{3} + \delta T_c \right)$$
$$= E_0 - V \left(\frac{\beta T_c^4}{12} + \frac{\delta T_c^2}{2} \right)$$

Solve for the gap Δ using $G_s(T_c) = G_n(T_c)$

$$\Rightarrow \Delta + \frac{\alpha T_c^4}{12} = \frac{\beta T_c^4}{12} + \frac{\delta T_c^2}{2}$$

$$\Delta = \frac{(\beta - \alpha) T_c^4}{12} + \frac{\delta T_c^2}{2}$$

With T_c from part (a),

$$\Delta = \frac{(\beta - \alpha)}{12} \left(\frac{3\delta}{\alpha - \beta} \right)^2 + \frac{\delta}{2} \left(\frac{3\delta}{\alpha - \beta} \right)$$

$$\Delta = \frac{3}{4} \frac{\delta^2}{\alpha - \beta} = \frac{1}{4} \delta T_c^2$$

e) In magnetic field B ,
$$\begin{cases} M_S = -\frac{VB}{4\pi} \\ M_n = 0 \end{cases}$$

$$dE = TdS + B dM + \mu dN, \quad N \text{ fixed } (dN=0)$$

In the presence of B ,

$$dG = -SdT - MdB + \mu dN$$

$$G_S(T, B) = E - TS - BM$$

$$= G_S(T, B=0) - \int_0^B M(B') dB'$$

$$= G_S(T, B=0) + \int_0^B \frac{VB'}{4\pi} dB'$$

$$= G_S(T, B=0) + \frac{VB^2}{8\pi}$$

$$= E_0 - V \left(\Delta + \frac{\alpha T^4}{12} \right) + \frac{B^2}{8\pi}$$

$$G_n(T, B) = G_n(T, B=0)$$

$$= E_0 - V \left(\frac{\beta T^4}{12} + \frac{\sigma}{2} T^2 \right)$$

Chemical Equilibrium: $G_S(T, B_c) = G_n(T)$

$$\rightarrow \frac{B_c^2}{8\pi} = \Delta - \frac{\sigma}{2} T^2 + \frac{\alpha - \beta}{12} T^4$$

$$\frac{B_c^2}{8\pi} = \frac{3}{4} \frac{\delta^2}{\alpha - \beta} - \frac{\delta}{2} T^2 + \frac{\alpha - \beta}{12} T^4$$

Δ from part d

$$= \frac{\alpha - \beta}{12} \left(\frac{3\delta}{\alpha - \beta} - T^2 \right)^2$$

$$= \frac{\alpha - \beta}{12} (T_c^2 - T^2)^2$$

$$= \frac{\alpha - \beta}{12} T_c^4 \left(1 - \frac{T^2}{T_c^2} \right)^2$$

$$= \frac{\alpha - \beta}{12} \left(\frac{3\delta}{\alpha - \beta} \right)^2 \left(1 - \frac{T^2}{T_c^2} \right)^2$$

$$= \frac{3}{4} \frac{\delta^2}{\alpha - \beta} \left(1 - \frac{T^2}{T_c^2} \right)^2$$

$$= \Delta \left(1 - \frac{T^2}{T_c^2} \right)^2$$

$$\Rightarrow B_c(T) = \sqrt{8\pi\Delta} \left(1 - \frac{T^2}{T_c^2} \right)$$

$$= \sqrt{8\pi \cdot \frac{1}{4} \delta T_c^2} \left(1 - \frac{T^2}{T_c^2} \right)$$

$$= T_c \sqrt{2\pi\delta} \left(1 - \frac{T^2}{T_c^2} \right)$$

$$\equiv B_0 \left(1 - \frac{T^2}{T_c^2} \right) \quad \text{with} \quad B_0 = \sqrt{8\pi\Delta} = \sqrt{2\pi\delta} T_c$$

Kardar, Chapter 2

$$1. a) p(x) = \begin{cases} \frac{1}{2a} & , -a < x < a \\ 0 & , \text{otherwise} \end{cases}$$

Characteristic function

$$\tilde{p}(k) = \int_{-\infty}^{\infty} p(x) e^{-ikx} dx$$

$$= \int_{-a}^a \frac{1}{2a} e^{-ikx} dx$$

$$= -\frac{1}{2a ik} e^{-ikx} \Big|_{-a}^a$$

$$\tilde{p}(k) = \frac{1}{ka} \sin(ka)$$

We can find the moments of $p(x)$ by expanding $\tilde{p}(k)$ in a Taylor series:

$$\tilde{p}(k) = \frac{1}{ka} \sum_{n=0}^{\infty} \frac{(-1)^n (ka)^{2n+1}}{(2n+1)!}$$

$$= \sum_{n=0}^{\infty} \frac{(-i ka)^{2n}}{(2n+1)!}$$

$$\langle x \rangle_c = \langle x \rangle = 0 \quad \text{from coeff. of } (-ik)^1$$

$$\langle x^2 \rangle_c = \langle x^2 \rangle - \langle x \rangle^2 = \langle x^2 \rangle = \frac{1}{3} a^2 \quad \text{from coeff. of } \frac{(-ik)^2}{2!}$$