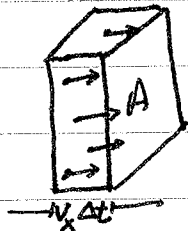


Phys 201 Problem Set 5 Solutions
F'09

3.4 $m_C \approx 12 m_H$, so $\boxed{12g}$ of C are required to mix with 1g of H to form C_2H_2 .

3.29 a)



All of the gas particles moving to the right within transverse distance $v_x \Delta t$ from the surface of area A pass through the surface in time Δt .

There are $n A v_x \Delta t$ particles in that volume, of which half move to the right.

Since $\frac{1}{2} m \langle v_x^2 + v_y^2 + v_z^2 \rangle = \frac{1}{2} m \cdot 3 \langle v_x^2 \rangle = \frac{3}{2} k_B T$,
 $v_x = \sqrt{\langle v_x^2 \rangle} = \sqrt{\frac{k_B T}{m}}$. We assume all of the particles have this speed.

Hence, $\frac{1}{2} n A v_x \Delta t = \boxed{\frac{n A \Delta t}{2} \sqrt{\frac{k_B T}{m}}}$ particles strike the surface.

b) $A = 1 \text{ cm}^2$, $T = 273 \text{ K}$, $\Delta t = 1 \text{ s}$, $k_B = 1.38 \times 10^{-23} \text{ J/K}$
 $= 10^{-4} \text{ m}^2$

$$m_{O_2} \approx 2 m_O = 2 \cdot (15.9995 \text{ u}) = 2 \cdot (15.9995) (1.66 \times 10^{-27} \text{ kg}) \\ = 5.31 \times 10^{-26} \text{ kg}$$

To calculate n use the ideal gas law!

$$PV = N k_B T \rightarrow n = \frac{N}{V} = \frac{P}{k_B T} = \frac{1.01 \times 10^5 \text{ N/m}^2}{(1.38 \times 10^{-23} \text{ J/K})(273 \text{ K})} \\ = 2.68 \times 10^{25} \text{ m}^{-3}$$

$$\frac{n A \Delta t}{2} \sqrt{\frac{k_B T}{m_{O_2}}} = (2.68 \times 10^{25} \text{ m}^{-3}) (10^{-4} \text{ m}^2) \frac{(1 \text{ s})}{2} \sqrt{\frac{(1.38 \times 10^{-23} \text{ J/K})(273 \text{ K})}{5.31 \times 10^{-26} \text{ kg}}} \\ = \boxed{3.57 \times 10^{23}} \text{ Oxygen molecules}$$

$$3.39 \quad D_{\text{rms}} = \sqrt{\frac{k_B T}{3\pi\eta a}} \sqrt{t} \Rightarrow t \propto D_{\text{rms}}^2 a$$

a = radius of object.

For $a = \frac{1}{2} \cdot (1 \mu\text{m})$, $D_{\text{rms}} = 5 \mu\text{m}$, $t = 20 \text{ s}$.

For $a = \frac{1}{2} \cdot 100 \mu\text{m}$, $D_{\text{rms}} = 500 \mu\text{m}$,

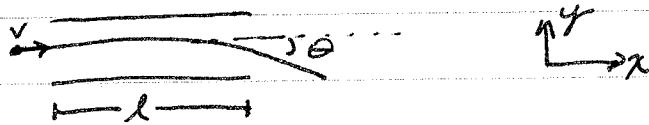
$$t = 20 \text{ s} \cdot \frac{(500 \mu\text{m})^2 (\frac{1}{2} \cdot 100 \mu\text{m})}{(5 \mu\text{m})^2 (\frac{1}{2} \cdot 1 \mu\text{m})}$$

$$= 20 \text{ s} \cdot 10^6$$

$$= \boxed{2 \times 10^7 \text{ s}}$$

This is much too long to have been observed.

3.40



Time spent by electron between plates: $\Delta t = \frac{l}{v}$

Acceleration: $\vec{a} = \frac{\vec{F}}{m} = -\frac{eE}{m} \hat{y}$

Velocity after passing plates: $\vec{v} = v \hat{x} - \frac{eE}{m} \Delta t \hat{y}$
 $= v \hat{x} - \frac{eE}{m} \cdot \frac{l}{v} \hat{y}$

$$v_x = v, \quad v_y = -\frac{eEl}{mv}$$

$$\tan \theta = \left| \frac{v_y}{v_x} \right| = \frac{eEl}{mv^2}$$

For small angles $\tan \theta \approx \theta \approx \boxed{\frac{eEl}{mv^2}}$

$$3.44 \quad K = \frac{1}{2} m v^2 = e V_0$$

$$R = \frac{m v}{e B} \Rightarrow v = \frac{R e B}{m}$$

$$\frac{1}{2} m \left(\frac{R e B}{m} \right)^2 = e V_0$$

$$= \frac{1}{2} R^2 B^2 \frac{e}{m} \cdot e$$

$$\rightarrow \boxed{\frac{m}{e} = \frac{R^2 B^2}{2 V_0}}$$