

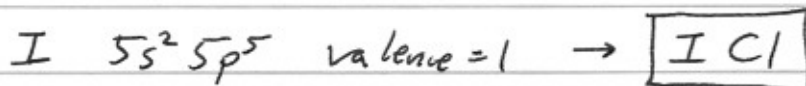
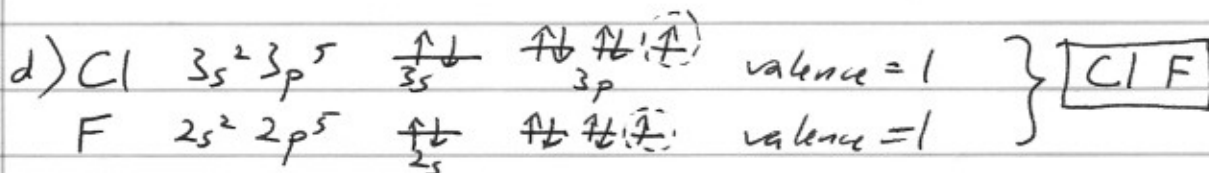
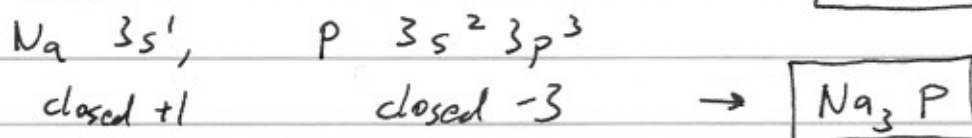
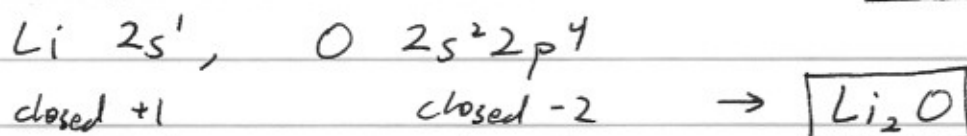
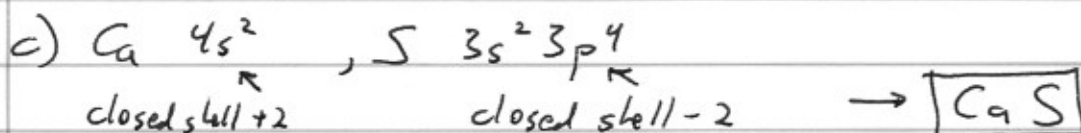
# Phys 201 F'08 Final Exam Solutions

1. a) After one electron is removed from the He atom, what is left is an  $\text{He}^+$  ion. Its ionization energy is

$$E_{\text{He}^+} = \boxed{+4 E_R} = \boxed{+54.4 \text{ eV}}$$

$\swarrow$   
 $z^2, \text{ with } z=2$

b) Binding energy =  $24.6 \text{ eV} + 54.4 \text{ eV} = \boxed{79.0 \text{ eV}}$



2. a)  $\text{Prob}(E = -E_R/4) = \frac{3^2}{10} = \boxed{\frac{9}{10}}$

b)  $\Psi(\vec{r}, t) = \frac{1}{\sqrt{10}} \left( \Psi_{100}(\vec{r}) e^{-iE_1 t/\hbar} + 3 \Psi_{211}(\vec{r}) e^{-iE_2 t/\hbar} \right)$

where  $E_1 = -E_R$ ,  $E_2 = -\frac{E_R}{4}$

c)  $-\frac{\hbar^2}{2m} \left( \frac{\partial^2 \Psi}{\partial x^2} + \frac{\partial^2 \Psi}{\partial y^2} + \frac{\partial^2 \Psi}{\partial z^2} \right) = E \Psi$

$\Psi(x, y, z) = \Psi_x(x) \Psi_y(y) \Psi_z(z)$

$-\frac{\hbar^2}{2m} \left( \frac{\Psi_x''(x)}{\Psi_x(x)} + \frac{\Psi_y''(y)}{\Psi_y(y)} + \frac{\Psi_z''(z)}{\Psi_z(z)} \right) = E$

fn. of  $x$                       fn. of  $y$                       fn. of  $z$                       ← independent constants

$-\frac{\hbar^2}{2m} \frac{\Psi_x''(x)}{\Psi_x(x)} = E_x, \quad -\frac{\hbar^2}{2m} \frac{\Psi_y''(y)}{\Psi_y(y)} = E_y, \quad -\frac{\hbar^2}{2m} \frac{\Psi_z''(z)}{\Psi_z(z)} = E_z$

with  $E_x + E_y + E_z = E$

d)  $\Psi_x(x) \propto \sin\left(\frac{n_x \pi x}{L}\right), \quad \Psi_y(y) \propto \sin\left(\frac{n_y \pi y}{L}\right), \quad \Psi_z(z) \propto \sin\left(\frac{n_z \pi z}{L}\right)$

$E_{n_x, n_y, n_z} = \frac{\hbar^2 \pi^2}{2mL^2} (n_x^2 + n_y^2 + n_z^2); \quad n_x, n_y, n_z \text{ integers } > 0.$

e) Ground state:  $n_x = n_y = n_z = 1 \rightarrow$  nondegenerate

1<sup>st</sup> excited state:  $(n_x, n_y, n_z) = (2, 1, 1)$  or  $(1, 2, 1)$  or  $(1, 1, 2)$

$\rightarrow$  degeneracy = 3

f) 5 electrons: By Pauli exclusion principle, 2 in ground state

w/  $E = \frac{3\hbar^2 \pi^2}{2mL^2}$ , 3 in 1<sup>st</sup> excited states w/  $E = \frac{3\hbar^2 \pi^2}{mL^2}$

$\rightarrow E = 2 \left( \frac{3\hbar^2 \pi^2}{2mL^2} \right) + 3 \left( \frac{3\hbar^2 \pi^2}{mL^2} \right) = \boxed{\frac{12\hbar^2 \pi^2}{mL^2}}$

3. a)  $u = \frac{pc^2}{E} = \frac{pc^2}{\sqrt{p^2c^2 + m^2c^4}} = c$  if  $m=0$ .

b)  $\vec{p}_M = \vec{p}_1 + \vec{p}_2 = (3\hat{x} + 4\hat{y}) \text{ GeV}/c$

$|\vec{p}_M| = 5 \text{ GeV}/c \equiv p_M$

$E_M = (p_1^2c^2 + m_1^2c^4)^{1/2} + (p_2^2c^2 + m_2^2c^4)^{1/2}$

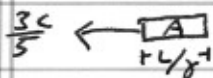
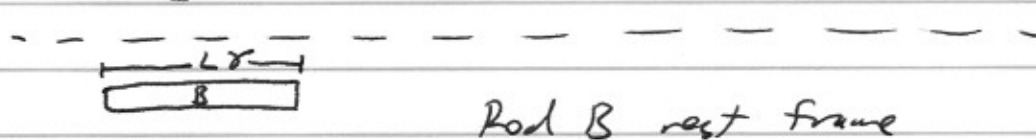
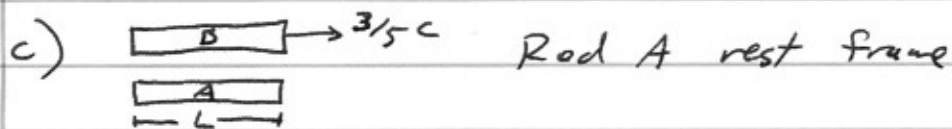
$= (16+20)^{1/2} \text{ GeV} + (9+16)^{1/2} \text{ GeV}$

$= 6 \text{ GeV} + 5 \text{ GeV} = 11 \text{ GeV}$

$p_M^2c^2 + M^2c^4 = 25(\text{GeV})^2 + M^2c^4 = 121(\text{GeV})^2$

$M = \sqrt{96} \text{ GeV}/c^2 = \boxed{4\sqrt{6} \text{ GeV}/c^2}$

$u_M = \frac{p_Mc^2}{E_M} = \frac{5 \text{ GeV} \cdot c}{11 \text{ GeV}} = \boxed{\frac{5}{11} c}$



$\frac{\text{Length of A}}{\text{Length of B}} = \frac{L/\gamma}{L\gamma} = \frac{1}{\gamma^2} = 1 - \frac{9}{25} = \boxed{\frac{16}{25}}$