

Wave-particle duality - for all!

We have discussed that ~~particles~~ light can behave either as a particle (photon) or a wave

Wave description

$f, \lambda, \text{electric field}$

$$\vec{E} = \vec{E}_0 \cos\left(\frac{2\pi}{\lambda} z - 2\pi f t + \varphi\right)$$

interference arises from coherent superposition of e-m field amplitudes

Particle description

momentum $p = h/\lambda$

energy $E = hf$

interference?

→ to reconcile these two paradigms we have to describe the light using a quantum wave function (probability wave)

A single photon with defined wavelength λ and p freq f

$$\psi(z, t) = \frac{1}{\sqrt{2\pi}} e^{ikz - i\omega t + i\varphi}$$

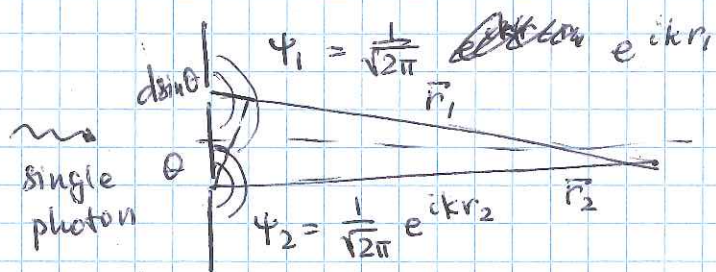
$k = \frac{2\pi}{\lambda}$ wave vector (spatial frequency)

$\omega = 2\pi f$ cyclic frequency

A probability of finding a photon at a particular outcome is determined by the absolute value squared of the wave function, so the measurable values are not sensitive to the wave function phase.

However, if the wave function has several components (i.e. photon can be in several states at once), the interference principle works for the wave functions, and the resulting outcome does depend on the relative phases b/w different paths.

Two-slit interference



Since we don't know which ~~the~~ slit the photon went through,

$$\psi_{\text{tot}} = \frac{1}{\sqrt{2}} (\psi_1 + \psi_2)$$

(50% chance - slit 1 $\left|\frac{1}{\sqrt{2}}\right|^2$
 - slit 2 $\left|\frac{1}{\sqrt{2}}\right|^2$)

Probability to find it anywhere at a screen

$$P = \frac{1}{2} |\psi_1 + \psi_2|^2 = \frac{1}{2} |e^{ikr}|^2 |1 + e^{ikd\sin\theta}|^2 = \cos^2\left[\frac{kd\sin\theta}{2}\right]$$

But what about other particles?

de Broglie idea - all particles have some wave properties. (pilot waves)

Particle's ~~kinetic~~ kinetic energy $E \rightarrow f = h/E$ (~~well~~)

Particle's momentum $p \rightarrow \lambda_p = \frac{h}{p} \approx \sqrt{\frac{2m}{E}}$

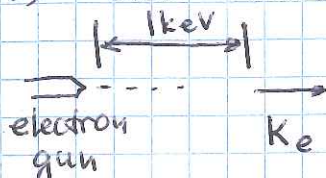
It is more common (and fun) to use $\hbar = h/2\pi$

$E = \hbar\omega$ $\hbar = 10^{-34}$ J.s de Broglie wavelength $\lambda_p = \frac{2\pi\hbar}{p} \Rightarrow k = \frac{p}{\hbar}$
 $p = \hbar k$

Why we don't see wave nature manifestation for most particles? λ_p is too small!

a) A Bowling ball $p = 1\text{kg} \cdot 1\text{m/s} \Rightarrow \lambda \approx 6 \cdot 10^{-34}\text{m}$

b) "Slow" electrons



 $K_e \approx 1\text{keV}$ (non-relativistic electrons)

 $K_e \approx \frac{p_e^2}{2m}$ $p_e \approx \sqrt{2m_e K_e} \approx \frac{2\pi\hbar}{\lambda_e} \approx 0.04\text{nm}$

 (comparable with X-rays)

↳ One can see the diffraction if e-beam reflects off ~~crystal~~ crystals.

c) Proton/neutrons - 2000 times heavier
at room temperature $K \approx 0.03\text{eV}$ ($T = 300\text{K}$)

$\lambda_p = \frac{2\pi\hbar}{\sqrt{2m_p K}} \approx 0.16\text{nm}$ → also can be used to probe small structure, including for bio/medical purposes!

Since electrons and protons are charged particles, they are often destructive for internal structures, but neutrons are very "gentle" (but one needs a reactor to produce them!)

Some helpful conversions for calculations

$$hc = 1240\text{ eV} \cdot \text{nm}$$
$$\lambda = \frac{hc}{\sqrt{2mk}} = \frac{hc}{\sqrt{2(mc^2)k}} \approx \frac{1240\text{ eV} \cdot \text{nm}}{\sqrt{2k}}$$

Electron: $K_e = 10^3\text{eV}$ $mc^2 = 0.5\text{MeV} = 5 \cdot 10^5\text{eV}$

$$\lambda_e = \frac{1240\text{ eV} \cdot \text{nm}}{\sqrt{2 \cdot 5 \cdot 10^5\text{eV} \cdot 10^3\text{eV}}} = \frac{1250\text{ eV} \cdot \text{nm}}{\sqrt{10^9\text{eV}}} = \frac{1250\text{ eV} \cdot \text{nm}}{3 \cdot 10^4\text{ eV}} \approx 0.04\text{nm}$$

How can we make the de Broglie wavelength larger? We need to make ~~tiny~~ particles colder!

Ultracold atoms

$$T \approx 10 \text{ nK} \Rightarrow K = 3 \cdot 10^{-12} \text{ eV}$$

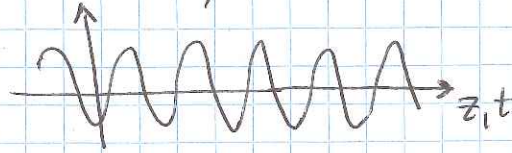
$$\lambda_p = \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{2 \cdot 10^9 \text{ eV} \cdot 3 \cdot 10^{-12} \text{ eV}}} = 1.6 \cdot 10^4 \text{ nm} = 16 \mu\text{m}$$

observable scale!

Uncertainty principle

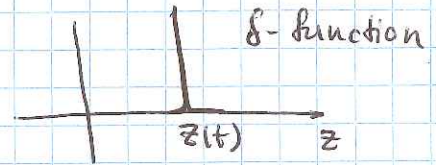
"Pure" wave — ~~can~~ exists uniformly in time and space

$$A \cos(kz - \omega t)$$



complete delocalization
well-defined momentum

"Pure" particle — absolute localization
well-defined position



wave vs particle
momentum vs position

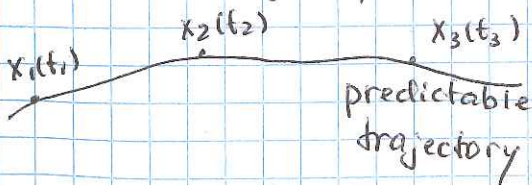
In classical physics we can know everything about any object, it is a completely deterministic description.

Since in quantum mechanics we are dealing with probabilistic ~~des~~ description, we may not be able to receive the complete information

Classical description
deterministic

measurements don't affect the state of a system, so consecutive measurements can provide all information about all parameters

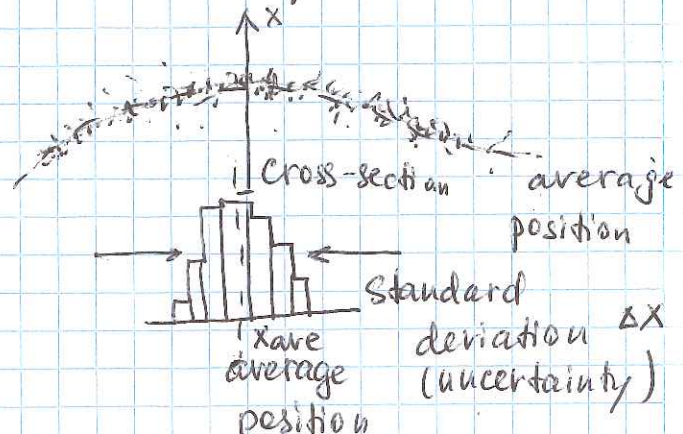
Classical trajectory



Quantum description
probabilistic

An act of measurement changes the state of a system, so only partial information can be obtained

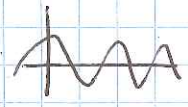
It is only possible to talk about average values and distributions



In quantum mechanics certain values cannot be measured simultaneously, i.e. a measurement of one parameter will change the outcome of a measurement ~~of~~ for another

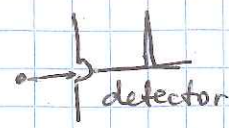
Position and momentum of a particle are such two parameters

Two extremes: wave



\vec{k} is precisely known
($\vec{p} = \hbar \vec{k}$)
no position information

particle



position is known
but momentum information is lost

In general, it is possible to determine average values for position and momentum and their uncertainty

position: $x_{ave} \pm \Delta x$ momentum $p_{ave} \pm \Delta p$

Heisenberg uncertainty principle

$$\Delta x \cdot \Delta p \geq \hbar$$

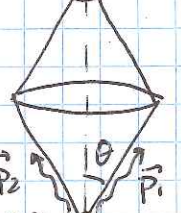
Fundamental limitation on measurement precision

Another important uncertainty principles

Energy - time $\Delta E \cdot \Delta t \geq \hbar$ (energy conservation can be violated for a short time!)
Photon number - phase $\Delta n \cdot \Delta \phi \geq 1$
(fundamental limit of optical measurements)

Physical interpretation of the uncertainty principle (Heisenberg microscope)

Δx



Limited optical resolution

$$\Delta x = \frac{\lambda}{2 \sin \theta}$$

However, in order for a photon to be scattered into the cone of the lens' vision, its momentum must change direction, and due to the momentum conservation, a momentum $\hbar k$ is given to the particle

$|\vec{p}_0| = |\vec{p}_1| = |\vec{p}_2| = \frac{2\pi\hbar}{\lambda}$

$$|\Delta \vec{p}_{photon}| = |\Delta \vec{p}_{part}| \approx 2 p_0 \sin \theta = 2 \frac{2\pi\hbar}{\lambda} \sin \theta$$

Thus, $\Delta x_{part} \cdot \Delta p_{part} \approx \frac{\lambda}{2 \sin \theta} \cdot \frac{2\pi\hbar}{\lambda} \cdot 2 \sin \theta \sim 2\pi\hbar$

$$\Delta x_{part} \Delta p_{part} \sim 2\pi\hbar$$

correct order-of-magnitude estimate

There is a rigorous mathematical basis to