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4.8

The Gibbs Canonical Ensemble

Suppose we allow the energy of a system to change by the addition of both heat and work from a reservoir.

$$\text{Macrostate } M = (T, \vec{J})$$

temperature \uparrow forces \leftarrow balanced w/ reservoir

(\vec{x}, E) $\mathcal{H}(\mu_s)$	(β, T) $\vec{J} \cdot \vec{x}$
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Energy and thermodynamic coordinates are random variables.

Energy of system + reservoir includes

$$\mathcal{H}(\mu_s) - \vec{J} \cdot \vec{x}$$

Energy change of reservoir is opposite the work done on the system.

$$\text{Probabilities } p(\mu_s, \vec{x}) = \frac{\exp[-\beta \mathcal{H}(\mu_s) + \beta \vec{J} \cdot \vec{x}]}{\mathcal{Z}(T, N, \vec{J})}$$

$$\text{where } \mathcal{Z}(T, N, \vec{J}) = \sum_{\mu_s, \vec{x}} \exp[-\beta \mathcal{H}(\mu_s) + \beta \vec{J} \cdot \vec{x}]$$

N fixed —
no chemical work

\mathcal{Z} = Gibbs partition function.

$$\langle \vec{x} \rangle = \sum_{\mu, \vec{x}} \vec{x} P(N, \mu, \vec{x})$$

$$= k_B T \frac{\partial \ln \mathcal{Z}}{\partial \vec{J}}$$

In terms of the Gibbs Free Energy

$$G(N, T, \vec{J}) = E - TS - \vec{x} \cdot \vec{J},$$

$$dG = -SdT - \vec{x} \cdot d\vec{J} + \mu dN \quad \text{fixed}$$

$$\vec{x} = - \frac{\partial G}{\partial \vec{J}}$$

Comparing with $\langle \vec{x} \rangle$, we identify

$$G(N, T, \vec{J}) = -k_B T \ln \mathcal{Z}$$

Enthalpy

$$H = E - \vec{x} \cdot \vec{J}$$

enthalpy \rightarrow

$$H = \langle \cancel{E} - \vec{x} \cdot \vec{J} \rangle = \frac{-\partial \ln \mathcal{Z}}{\partial \beta}$$

\swarrow Hamiltonian

$$\text{Heat Capacities } C_{\vec{J}} = \left. \frac{\partial H}{\partial T} \right|_{\vec{J}}$$

Example: Ideal Gas - Isobaric ensemble
(fixed pressure)

Macrostate $M = (T, N, P)$

Microstate $\mu = \{ \vec{p}_i, \vec{q}_i \}$, volume V

↑
random variable

$$P(\{ \vec{p}_i, \vec{q}_i \}, V) = \frac{1}{Z} \exp \left[-\beta \sum_{i=1}^N \frac{p_i^2}{2m} - \beta P V \right]$$

$\times \begin{cases} 1 & \text{for } \{ \vec{q}_i \} \in V \\ 0 & \text{otherwise.} \end{cases}$

Cubic Partition Function:

$$Z(N, T, P) = \int_0^\infty dV \left(e^{-\beta P V} \int \frac{1}{N!} \prod_{i=1}^N \frac{d^2 q_i d^3 p_i}{h^3} \exp \left[-\beta \sum_{i=1}^N \frac{p_i^2}{2m} \right] \right)$$

↑ identical particles

$$= \left(\int_0^\infty V^N e^{-\beta P V} \right) \cdot \frac{1}{N! \lambda(T)^{3N}}$$

$$Z(N, T, P) = \frac{1}{(\beta P)^{N+1} \lambda(T)^{3N}}$$

$$\stackrel{N \gg 1}{\approx} \frac{1}{P^N} \left(\frac{2\pi m}{h^2} \right)^{3N/2} (k_B T)^{5N/2}$$

where $\lambda(T) = \frac{h}{\sqrt{2\pi m k_B T}}$

36s Free Energy:

$$G = -k_B T \ln Z \approx N k_B T \left[\ln p - \frac{5}{2} \ln(k_B T) + \frac{3}{2} \ln \left(\frac{h^2}{2\pi m} \right) \right]$$

$$dG = -SdT + VdP + \mu dN$$

$$\text{volume } V = \left. \frac{\partial G}{\partial P} \right|_{T, N} = \frac{N k_B T}{P}$$

$$\rightarrow PV = N k_B T$$

Entropy:

$$H = \langle E + PV \rangle = - \left. \frac{\partial \ln Z}{\partial \beta} \right|_{P, N} = \frac{5}{2} N k_B T$$

Specific heat at constant pressure:

$$C_P = \frac{dH}{dT} = \frac{5}{2} N k_B$$

Chemical potential:

$$\mu = \left. \frac{\partial G}{\partial N} \right|_{P, T} = k_B T \ln \left(\frac{P \lambda(T)^3}{k_B T} \right)$$

Example: Spins in an external magnetic field

Work done against the magnetic field: $-\vec{B} \cdot \vec{M}$

magnetic field \nearrow
magnetization \uparrow

Macrostate $M = (N, T, \vec{M})$

Microstate $\mu = \{\text{Spins about } \vec{B}: \sigma_i = \pm 1\}$

Magnetization about \vec{B} : $M = \mu_0 \sum_{i=1}^N \sigma_i$

\nwarrow microscopic magnetic moment.

$$\mathcal{Z}(N, T, \vec{B}) = \sum_{\{\sigma_i = \pm 1\}} \exp(\beta \vec{B} \cdot \vec{M})$$

$$= \sum_{\{\sigma_i = \pm 1\}} \exp\left(\beta B \mu_0 \sum_{i=1}^N \sigma_i\right)$$

$$\mathcal{Z} = (2 \cosh(\beta \mu_0 B))^N$$

$$\text{probability } p(\{\sigma_i\}) = \frac{\exp\left[\beta B \mu_0 \sum_{i=1}^N \sigma_i\right]}{\mathcal{Z}}$$

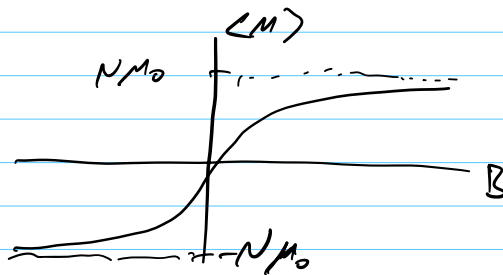
\mathcal{Z}

Gibbs Free Energy:

$$G = -k_B T \ln Z = -N k_B T \ln [2 \cosh(\beta B \mu_0)]$$

Magnetization: $dG = -SdT - MdB + \mu dN$

$$\langle M \rangle = - \left. \frac{\partial G}{\partial B} \right|_{T, N} = N \mu_0 \tanh(\beta \mu_0 B)$$



Magnetic Susceptibility:

$$\chi(T) = \left. \frac{\partial M}{\partial B} \right|_{T, N} = N \mu_0^2 \left(\operatorname{sech}(\beta \mu_0 B) \right)^2 \Big|_{B \rightarrow 0}$$

$$\Rightarrow \boxed{\chi(T) = \frac{N \mu_0^2}{k_B T}} \quad \text{Curie's law}$$

$$\text{Entropy } H = \langle \mathcal{H} - \beta M \rangle = -\beta M$$

Heat capacity at const. B :

$$C_B = \frac{dH}{dT} = -B \frac{\partial M}{\partial T}$$

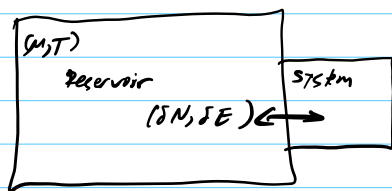
$$= \frac{N \mu_0^2 B}{k_B T^2} \left(\text{sech}(\beta \mu_0 B) \right)^2$$

Grand Canonical Ensemble

If we allow chemical work instead of mechanical work, and heat transfer from the reservoir, then:

Macrostate $M = (T, \mu, \vec{x})$
↳ chemical potential

Number of particles N is a random variable



Prob of system microstate μ_s :

$$P(\mu_s) = \frac{\exp[\beta \mu N(\mu_s) - \beta \mathcal{H}(\mu_s)]}{Q(T, \mu, \vec{x})}$$

$$\text{Grand Partition Function } Q(T, \mu, \vec{x}) = \sum_{\mu_s} \exp[\beta \mu N(\mu_s) - \beta \mathcal{H}(\mu_s)]$$

$$= \sum_{N=0}^{\infty} \left(e^{\beta \mu N} \sum_{\mu_s | N} e^{-\beta \mathcal{H}(\mu_s)} \right)$$

↑
sum over N .

↳ sum over microstates w/ fixed N

$$= \sum_{N=0}^{\infty} e^{\beta \mu N} Z(T, N, \vec{x})$$

$$\text{prob. } p(N) = \frac{e^{\beta \mu N} Z(T, N, \vec{x})}{Q(T, \mu, \vec{x})}$$

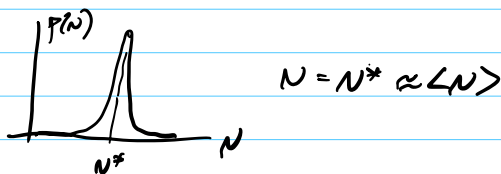
Average number of particles?

$$\langle N \rangle = \sum N p(N)$$
$$= \frac{1}{Q} \frac{\partial}{\partial (\beta \mu)} Q = \frac{\partial}{\partial (\beta \mu)} \ln Q$$

Number fluctuations:

$$\langle N^2 \rangle_c = \langle N^2 \rangle - \langle N \rangle^2$$
$$= \frac{1}{Q} \frac{\partial^2}{\partial (\beta \mu)^2} Q - \left(\frac{\partial}{\partial (\beta \mu)} \ln Q \right)^2$$
$$= \frac{\partial^2}{\partial (\beta \mu)^2} \ln Q$$
$$\approx \frac{\partial \langle N \rangle}{\partial (\beta \mu)} \approx N$$

$$\frac{\sqrt{\langle N^2 \rangle_c}}{\langle N \rangle} \rightarrow 0 \text{ as } N \rightarrow \infty$$



Then,

$$Q(T, \mu, \bar{x}) = \sum_{N \geq 0} e^{\beta \mu N} Z(T, \mu, \bar{x})$$
$$\approx e^{\beta \mu N^*} Z(T, N^*, \bar{x})$$
$$= e^{\beta \mu N^* - \beta F} = e^{-A(-\mu N^* + E - TS)}$$
$$= e^{-\beta \mathcal{G}}$$

\mathcal{G} Grand potential

$$\text{where } \mathcal{H}(T, \mu, \bar{x}) = E - TS - \mu N \\ = -k_B T \ln Q$$

$$d\mathcal{H} = -SdT + \bar{J} \cdot d\bar{x} - N d\mu$$

$$\Rightarrow S = -\left. \frac{\partial \mathcal{H}}{\partial T} \right|_{\bar{x}, \mu}, \quad N = -\left. \frac{\partial \mathcal{H}}{\partial \mu} \right|_{\bar{x}, T}, \quad \bar{J}_i = \left. \frac{\partial \mathcal{H}}{\partial x_i} \right|_{\mu, T, x_{j \neq i}}$$

Example: Ideal Gas

Macrostate $M = (T, \mu, V)$

Microstate $\mu = \{\vec{p}_i, \vec{q}_i\}$, with indefinite particle number

$$Q(T, \mu, V) = \sum_{N=0}^{\infty} e^{\beta \mu N} \frac{1}{N!} \int \prod_{i=1}^N \frac{d^3 p_i d^3 q_i}{h^3} \exp\left[-\beta \frac{p_i^2}{2m}\right] \\ = \sum_{N=0}^{\infty} e^{\beta \mu N} \cdot \frac{1}{N!} \left(\frac{V}{\lambda^3}\right)^N, \quad \text{with } \lambda = \frac{h}{\sqrt{2\pi m k_B T}} \\ = \exp\left[e^{\beta \mu} \frac{V}{\lambda^3}\right]$$

$$\mathcal{H}(T, \mu, V) = -k_B T \ln Q = -k_B T e^{\beta \mu} \frac{V}{\lambda^3}$$

$$\mathcal{H} = E - TS - \mu N = -PV$$

$$\Rightarrow P = -\frac{\mathcal{H}}{V} = k_B T \frac{e^{\beta \mu}}{\lambda^3}$$

$$PV = N k_B T$$

Particle Number:

$$N = -\left. \frac{\partial \mathcal{H}}{\partial \mu} \right|_{T, V} = e^{\beta \mu} \frac{V}{\lambda^3}$$

Chemical potential

$$\mu = k_B T \ln\left(\lambda^3 \frac{N}{V}\right) = k_B T \ln\left(\frac{P \lambda^3}{k_B T}\right)$$

so we for μ